

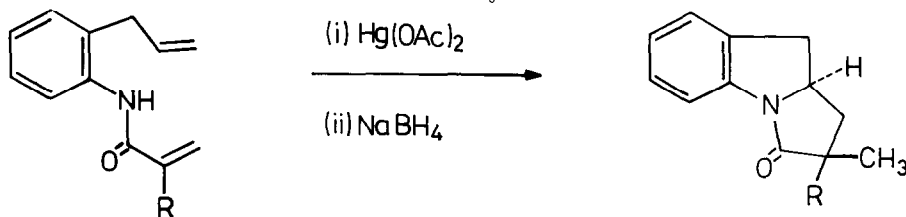
CYCLIZATIONS OF MERCURY AND PALLADIUM SUBSTITUTED ACRYLANILIDES

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Summary: Cyclizations via reductive coupling of a mercurial acrylanilide and carbo-
 metallation of a palladium derivative, each formed via amidometallation,
 follow a similar course.

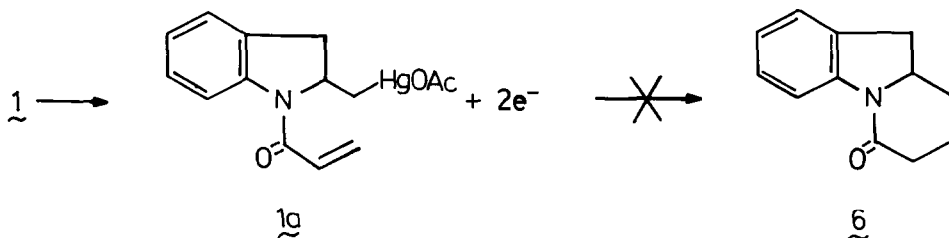
Ureidomercuration of N-Cbz-o-allylaniline followed by intermolecular reductive coupling
 with acrylonitrile, de-protection and cyclization provides a route to the benzoinolizidinone
 $\mathfrak{6}$.¹ In the light of our findings pertaining to the intramolecular reductive coupling of mer-
 curials,² a more convergent possibility, implied in the sequence $\mathfrak{1} \rightarrow \mathfrak{1a} \rightarrow \mathfrak{6}$, presented itself.

Accordingly, the acrylanilide $\mathfrak{1}$ was treated with mercuric acetate in the usual way.¹
 T.l.c. analysis indicated the disappearance of starting material. The presumed $\mathfrak{1a}$ was treated
 with sodium borohydride.³ Chromatography of the resultant product on silica gel afforded a 34%
 yield of $\mathfrak{3}$ ⁴ R= α H, δ CH₃ 1.25, dJ=7Hz(3) and an 11% yield of $\mathfrak{4}$ ⁴, R= β H; δ CH₃ 1.35, d, J=7Hz(3). A
 single product $\mathfrak{5}$ ⁴ (δ 1.25, dJ=7Hz(3), δ 1.33, dJ=7Hz(3)) was obtained from a similar reaction
 sequence on substrate $\mathfrak{2}$. In this case, a 28% yield of recovered starting material was isolated.
 Since disappearance of starting material was indicated at the mercuriation stage, reductive
 retro-amidomercuration is apparently competitive with reductive cyclization. In neither case
 was there any indication for the formation of compound $\mathfrak{6}$.



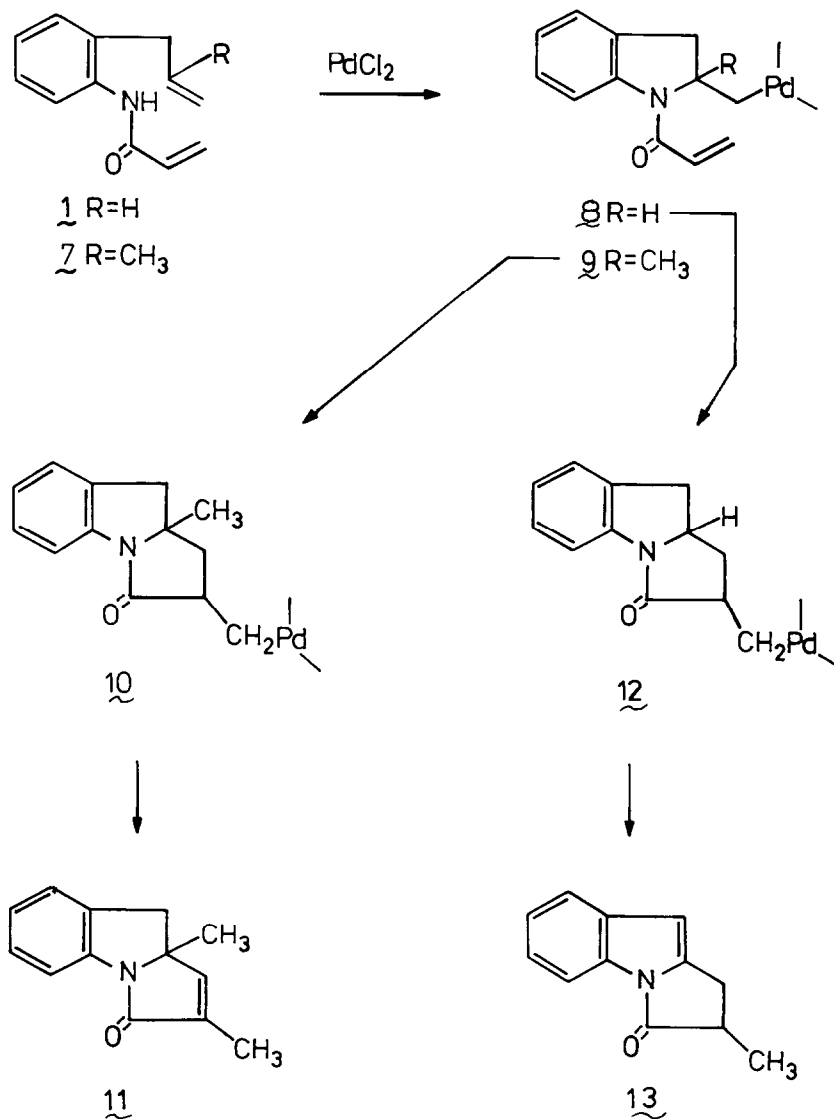
$\mathfrak{1}$ R=H
 $\mathfrak{2}$ R=CH₃

$\mathfrak{3}$ R = α -H
 $\mathfrak{4}$ R = β -H
 $\mathfrak{5}$ R = CH₃



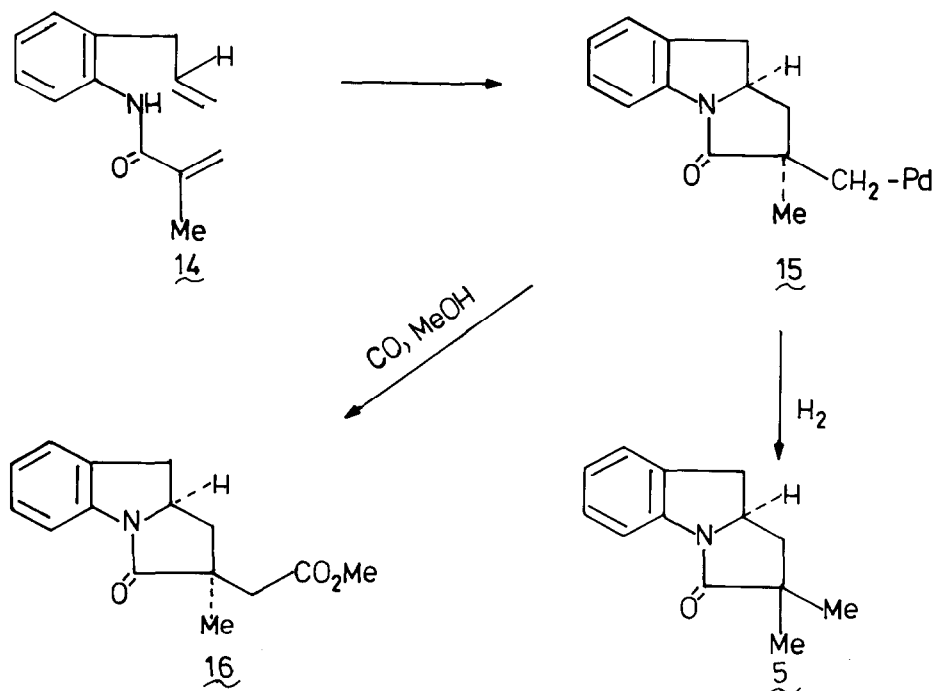
Cyclization to produce the 5- rather than the 6- membered ring was reminiscent of the interesting findings of Hegedus,^{5,6} wherein compound 7 suffers palladium mediated cyclization to afford 11 presumably via intermediates 9 and 10. Of course, with substrate 7 β -hydride elimination is precluded at the stage of 9 by the quaternary carbon (R=Me).

To probe more closely the degree of similarity between the two processes, acrylanilide 1 was subjected to amidopalladation. Reaction of 1 with $\text{PdCl}_2 (\text{O} \text{C}=\text{N})_2$ in tetrahydrofuran for 23 hr. at rt. afforded a 55% yield of 13.⁴ Presumably, this compound arises from retro-hydrido and hydrido-palladations starting with intermediate 12. Thus, carbopalladation (8 \rightarrow 12) is easily competitive with β -hydride elimination - a fact not implicit in the Hegedus experiment.⁵



Since the sequence from the postulated $\underline{12}$ to the observed $\underline{13}$ is rather complicated, additional support for the proposed pathway was sought. Methacrylanilide $\underline{14}$ was subjected to the same process as $\underline{1}$. It was recognized that the β -quarternary carbon, in what would be the corresponding intermediate $\underline{15}$, would preclude the retro-hydridopalladation step required for indole formation.⁷ The system, resulting from treatment of $\underline{14}$ with $\text{PdCl}_2 (\text{O}=\text{N})_2$ in tetrahydrofuran at rt. for 23 hr., was treated with hydrogen. There was obtained a 23% yield of the previously encountered $\underline{5}$. More remarkably, when the palladation product was treated with CO-methanol-triethylamine, there was obtained, albeit only in 5% yield, the tricyclic methyl ester, $\underline{16}$.^{4,8}

In the palladium mediated reaction the proposed sequence of amidometallation to bicyclic product followed by carbometallation to afford a 5-membered ring is thus strongly supported. The formation of the corresponding 5-membered ring in the mercury case (cf. $\underline{1a} \rightarrow \underline{3}$) may involve carbometallation or might best be viewed in terms of a "metal-free" radical cyclization. Experiments intended to address questions of this kind, as well as to clarify more fully the synthetic potentialities of such transformations, will be undertaken.



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References

1. S. Danishefsky and E. Taniyama, Tetrahedron Lett., preceding paper
2. S. Danishefsky, S. Chackalamannil and B. J. Uang, J. Org. Chem., 47, 2531 (1982)
3. The intermolecular version of this reaction was developed by B. Giese and associates. See: B. Giese, H. Horler and W. Zwick, Tetrahedron Lett. 931 (1982); B. Giese and K. Heuck, Chem. Ber. 114, 1572 (1981).
4. The structure of this compound is consistent with its infrared nmr and mass spectral properties.
5. L. S. Hegedus, G. F. Allen and D. J. Olsen, J. Amer. Chem. Soc., 102, 3583 (1980)
For a more recent variation of this reaction see: L. S. Hegedus and J. M. McKearin, ibid, 104, 2445 (1982).
6. That the cyclization of the mercurial occurs in the "anti-Michael" sense is no doubt related to the radical nature of the process, cf. C. L. Hill and G. M. Whitsides, ibid, 96, 87 (1974).
7. In the Hegedus work,⁵ at higher temperature, a palladium specie corresponding to 15 undergoes a most interesting fragmentation re-combination sequence, leading to a 6-membered ring.
8. Only one stereoisomer was noted. In view of the poor yield the significance of this finding cannot be evaluated. The stereochemical assignment for compound 16 is based on a comparison of its nmr spectrum with those of compounds 3, 4 and 5. Full spectral details will be published in a full paper.

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